SECTION 1: Identification of the substance/mixture and of the company/undertaking

1.1. Product identifier

Product name          HYDROCHLORIC ACID 0.1M
Product number        1045

1.2. Relevant identified uses of the substance or mixture and uses advised against

Identified uses          General chemical reagent
Uses advised against    Processes involving alkaline substances or cyanide compounds. These may produce an exothermic reaction or hydrogen cyanide gas.

1.3. Details of the supplier of the safety data sheet

Supplier
Reagent Chemical Services
18 Aston Fields Road
Whitehouse Industrial Estate
Runcorn
Cheshire WA7 3DL

T: 01928 716903 (08.30 - 17.00)
F: 01928 716425
E: info@reagent.co.uk

1.4. Emergency telephone number

Emergency telephone
OHES Environmental Ltd 24-7
Tel. 0333 333 9939 (24 hour)

SECTION 2: Hazards identification

2.1. Classification of the substance or mixture

Classification (EC 1272/2008)

Health hazards            Not Classified
Environmental hazards     Not Classified

Human health             Not classified but may cause temporary eye or skin irritation.
Environmental            The product is miscible with water and can spread in water systems. Although not classified as harmful to the environment the material should not be discharged to land or water systems, this may have an impact on the organisms in the local area. The product may produce a local pH change in water systems which can have an effect on aquatic organisms.
Physicochemical          May produce an exothermic reaction with alcalis, oxidising agents or other acids. May corrode metals. May react violently with alkali and alkali earth metals. May produce hydrogen gas on reaction with metals.
HYDROCHLORIC ACID 0.1M

Pictogram

Signal word Warning

Hazard statements H290 May be corrosive to metals.

Precautionary statements P234 Keep only in original container.
P390 Absorb spillage to prevent material damage.
P406 Store in corrosive resistant container.

2.3. Other hazards

This substance is not classified as PBT or vPvB according to current EU criteria.

SECTION 3: Composition/information on ingredients

3.2. Mixtures

<table>
<thead>
<tr>
<th>HYDROCHLORIC ACID ...%</th>
<th>5-10%</th>
</tr>
</thead>
<tbody>
<tr>
<td>CAS number: 7647-01-0</td>
<td></td>
</tr>
<tr>
<td>EC number: 231-595-7</td>
<td></td>
</tr>
<tr>
<td>REACH registration number: 01-211948462-27-0000</td>
<td></td>
</tr>
</tbody>
</table>

Classification
Skin Corr. 1B - H314
STOT SE 3 - H335

Classification (67/548/EEC or 1999/45/EC) C;R34 Xi;R37

The Full Text for all R-Phrases and Hazard Statements are Displayed in Section 16.

SECTION 4: First aid measures

4.1. Description of first aid measures

General information CAUTION! First aid personnel must be aware of own risk during rescue! Always consider any dangers in the vicinity before approaching to treat the casualty. First aid personnel must protect themselves with all necessary personal protective equipment during the assistance of casualties. When breathing is difficult, properly trained personnel may assist the casualty by administering oxygen. Check airway for any blockages. Place unconscious person on the side in the recovery position and ensure breathing can take place. Never give anything by mouth to an unconscious person. If breathing has stopped perform CPR. If medical assistance is needed take as much detail as possible about the incident and hazardous materials involved with the casualty.

Inhalation Remove affected person from source of contamination. Move affected person to fresh air and keep warm and at rest in a position comfortable for breathing. Get medical attention if any discomfort continues.

Ingestion Do not induce vomiting. Rinse mouth thoroughly with water. In case of ingestion of large amounts or if any discomfort continues obtain medical attention.

Skin contact As a general precaution remove contaminated clothing and wash the skin with plenty of water. If irritation or discomfort occurs obtain medical attention.

Eye contact Promptly wash eyes with plenty of water or eye wash solution while lifting the eyelids. If possible remove any contact lenses and continue to wash. Get medical attention if any discomfort continues.

4.2. Most important symptoms and effects, both acute and delayed
HYDROCHLORIC ACID 0.1M

General information
The severity of the symptoms described will vary dependent on the concentration and the length of exposure.

Inhalation
High concentrations of vapours may irritate the respiratory system.

Ingestion
Small amounts will leave taste in mouth, larger amounts may cause nausea or vomiting. May irritate the mouth and throat. Larger amounts may irritate the digestive system.

Skin contact
May irritate the skin.

Eye contact
May cause eye irritation.

4.3. Indication of any immediate medical attention and special treatment needed

Notes for the doctor
Have facilities in place to wash skin and eyes in case of exposure.

SECTION 5: Firefighting measures

5.1. Extinguishing media
Suitable extinguishing media
The product is non-combustible. Use fire-extinguishing media suitable for the surrounding fire. Water spray, foam, dry powder or carbon dioxide.

Unsuitable extinguishing media
Do not use water jet as this can spread the fire. Do not use carbon dioxide in enclosed spaces with insufficient ventilation.

5.2. Special hazards arising from the substance or mixture
Specific hazards
In case of fire, toxic or irritating fumes or vapours may be formed. No hazards associated with the product. Product containers are likely to melt in the heat of a fire.

Hazardous combustion products
Chlorine compounds. Hydrogen chloride (HCl).

5.3. Advice for firefighters
Protective actions during firefighting
Prevent run-off from entering drains and watercourses.

Special protective equipment for firefighters
Wear positive-pressure self-contained breathing apparatus (SCBA) and appropriate protective clothing.

SECTION 6: Accidental release measures

6.1. Personal precautions, protective equipment and emergency procedures
Personal precautions
The following is given as general advice, precautions and procedures should reflect the extent of a spillage and the situation. Use protective clothing and equipment as described in section 8 of this datasheet. Avoid ingestion, inhalation of vapours and contact with skin and eyes. Provide adequate ventilation. Use suitable respiratory equipment if spillages occur in enclosed spaces and vapours are produced. Have emergency procedures in place for treating spillages, evacuating the area and informing the emergency services if necessary. Restrict access to the area until the spillage is treated, if large amounts of vapours are produced that will be hazardous to others, evacuate the area. When any other effects of spillages will affect the safety of others the area should be evacuated.

6.2. Environmental precautions
HYDROCHLORIC ACID 0.1M

Environmental precautions
Although not classified as environmentally hazardous the mixture is acidic which can have an effect on pH. Avoid unauthorised discharge to the environment. Do not discharge into drains or watercourses or onto the ground. Clean up any spillages immediately, prevent material from spreading and entering drains or sewage systems. If spillages to land cannot be treated safely or if contamination will occur the Environment Agency must be alerted immediately. Large spillages or uncontrolled discharge to water systems must be alerted to the Environmental Agency or other regulatory body. If the substance has entered a foul drain or sewage system in significant quantity to cause a hazard the local Water Treatment Company must be informed.

6.3. Methods and material for containment and cleaning up

Methods for cleaning up
The method for cleaning spillages will be dependent upon the size of the spillage, unless amounts are very small and will be no risk to the environment it is advisable not to flush to drain due to the acidic pH. If in doubt, consult with the local authority regarding discharge to drain. Small Spillages: Absorb with inert, non-combustible material. Large Spillages: Dam and absorb spillages with sand, earth or other inert, non-combustible material. Fit drain covers where they are available if the spillage is likely to enter the drainage system. Collect and place in suitable waste disposal containers and seal securely. For waste disposal, see Section 13. Containers with collected spillage must be properly labelled with correct contents and hazard symbol. Wash spillage site well with water and detergent, be aware of the potential for surfaces to become slippery. Ventilate area and allow to dry before allowing access. Wash thoroughly after dealing with a spillage.

6.4. Reference to other sections

Reference to other sections
Refer to sections 8 and 13 for additional information.

SECTION 7: Handling and storage

7.1. Precautions for safe handling

Usage precautions
Avoid spilling. Avoid contact with skin and eyes. Avoid inhalation of vapours and spray/mists. Avoid ingestion of the product. Do not eat, drink or smoke when handling. Do not mix with incompatible substances or mixtures. Do not use in areas close to drainage systems unless measures are in place to prevent access of product. Wash at the end of each work shift and before using the toilet. Remove contaminated clothing/footwear/equipment before entering eating areas or other places that would expose others to the substance. Do not dispose of the substance to the environment through unauthorised means. Do not discharge to land or water including the drainage system. Ensure emergency procedures are in place to treat spillages and cope with other situations such as evacuation.

7.2. Conditions for safe storage, including any incompatibilities

Storage precautions
Store in area with adequate ventilation and sufficient air movement to prevent any build up of vapours. Store in closed original container at temperatures between 15°C and 25°C. Store away from heat, direct sunlight and moisture. Store away from incompatible materials. Keep above the chemical's freezing point. Store in a stable situation to avoid spillages. It is advisable to store in a bunded area or use other protective measures such as a sump pallet or storage tray. If the substance is transferred to other containers ensure the packaging material is compatible. Consult with the packaging manufacturer or supplier. Do not leave storage containers exposed to the atmosphere as this may result in loss of contents or contamination.

7.3. Specific end use(s)

Specific end use(s)
The identified uses for this product are detailed in Section 1.2.

SECTION 8: Exposure Controls/personal protection

8.1. Control parameters

Occupational exposure limits
HYDROCHLORIC ACID 0.1M

**HYDROCHLORIC ACID ...%**

Long-term exposure limit (8-hour TWA): WEL 1 ppm 2 mg/m³ gas and aerosol mists
Short-term exposure limit (15-minute): WEL 5 ppm 8 mg/m³ gas and aerosol mists
WEL = Workplace Exposure Limit

**Ingredient comments**
The following information refers to hydrogen chloride as a substance not the product as a mixture.

**DNEL**
- Inhalation; Short term local effects: 15 mg/m³
  Taken from the ECHA website: List of Registered Substances - Toxicity data.
- Inhalation; Long term local effects: 8 mg/m³

**PNEC**
- Fresh water; 36 µg/L
  Taken from the ECHA website: List of Registered Substances - Ecotoxicity data.
- Marine water; 36 µg/L
- Intermittent release; 45 µg/L
- STP; 36 µg/L

**HYDROCHLORIC ACID ...% (CAS: 7647-01-0)**

**DNEL**
Workers - Inhalation; Long term local effects: 8 mg/m³
Workers - Inhalation; Short term local effects: 15 mg/m³

### 8.2. Exposure controls

**Appropriate engineering controls**
Provide adequate general and local exhaust ventilation.

**Eye/face protection**
Wear approved chemical safety goggles conforming to EN 166.

**Hand protection**
Use full length gloves. Butyl rubber. Polyvinyl chloride (PVC). Nitrile rubber. Viton rubber (fluoro rubber). For gloves involving total immersion 1.0mm thickness (if available) are recommended, at least 0.5mm and breakthrough time of >480 minutes. For splash resistance use minimum 0.5mm thickness and breakthrough time > 240 minutes. Gloves should have a breakthrough time sufficient for the amount of handling but allow dexterity for safe movement and handling. Gloves showing signs of degradation should be changed to avoid skin contamination. Gloves should carry the CE mark and conform to BS EN 374, chemicals and micro-organisms. The most suitable glove should be chosen in consultation with the glove supplier/manufacturer, who can provide information about the breakthrough time of the glove material. When packages of the product are being handled during storage or transport it is advisable to wear protective gloves to prevent damage to the skin.

**Other skin and body protection**
Wear suitable protective clothing during transport, handling and storage operations connected with the product. Protective clothing should conform to the general requirements of EN 340:2003. Also consider EN 13034:2005; EN 14605:2005; EN 943:2002 dependent upon the situation resulting in exposure. Wear suitable protective footwear during handling of the product. When treating spillages it is recommended to wear protective boots, consult with the supplier as to the compatibility. Safety footwear should conform to standards EN 344 - 347. Wear plastic apron and full length gloves if handling large amounts. If there is a risk of splashing then wear a face shield. Have facilities in place to wash eyes in case of contact. If handling large amounts it is recommended to have a safety shower.

**Hygiene measures**
Remove clothing when contamination will result in exposure to the substance, segregate and wash before re-use. Do not eat, drink or smoke in the work area. Wash at the end of each work shift and before eating, smoking and using the toilet. Remove contaminated clothing when entering eating areas or other places that could lead to contamination of others with the product.
HYDROCHLORIC ACID 0.1M

Respiratory protection

Wear suitable respiratory protection when vapours or mists are produced if the Workplace Exposure Limit is exceeded and there is insufficient ventilation or extraction. When vapours are generated during spill clean up operations and exposure of operators is likely then respiratory equipment should be worn. Use respirator fitted with a cartridge suitable for inorganic acids including hydrochloric acid. When the concentration of acid vapours in the atmosphere is sufficient to cause skin irritation then wear a full face respirator. CAUTION: Air purifying respirators do not protect the user in oxygen deficient atmospheres, use air supplied system. Respiratory protection should conform to the following standards. BS EN 140: Half-face masks. BS EN 136: Full face masks. Powered air respirators should meet requirements of EN146 and EN12941. Airline fed respirators should meet the requirements of EN 270 and EN1835. Respiratory protection should be maintained in a proper condition and inspected at the frequency specified by current legislation.

Environmental exposure controls

See section 6 for details.

SECTION 9: Physical and Chemical Properties

9.1. Information on basic physical and chemical properties

<table>
<thead>
<tr>
<th>Property</th>
<th>Value/Note</th>
</tr>
</thead>
<tbody>
<tr>
<td>Appearance</td>
<td>Liquid.</td>
</tr>
<tr>
<td>Colour</td>
<td>Colourless.</td>
</tr>
<tr>
<td>Odour</td>
<td>Pungent.</td>
</tr>
<tr>
<td>pH (concentrated solution)</td>
<td>&lt;1 Not determined. pH (diluted solution): The value will depend on the concentration of the diluted solution but will remain &lt;7</td>
</tr>
<tr>
<td>Melting point</td>
<td>Not determined.</td>
</tr>
<tr>
<td>Initial boiling point and range</td>
<td>Not determined.</td>
</tr>
<tr>
<td>Flash point</td>
<td>Not relevant. The mixture is non-flammable.</td>
</tr>
<tr>
<td>Evaporation rate</td>
<td>Not determined.</td>
</tr>
<tr>
<td>Evaporation factor</td>
<td>Not determined.</td>
</tr>
<tr>
<td>Flammability (solid, gas)</td>
<td>No.</td>
</tr>
<tr>
<td>Upper/lower flammability or</td>
<td>Not relevant.</td>
</tr>
<tr>
<td>explosive limits</td>
<td></td>
</tr>
<tr>
<td>Vapour pressure</td>
<td>Not determined.</td>
</tr>
<tr>
<td>Vapour density</td>
<td>Not determined.</td>
</tr>
<tr>
<td>Relative density</td>
<td>Approx. 1.0 - 1.05 @ @ 20°C</td>
</tr>
<tr>
<td>Bulk density</td>
<td>Not relevant.</td>
</tr>
<tr>
<td>Solubility(ies)</td>
<td>Not determined. Miscible with water. The product is completely miscible with water.</td>
</tr>
<tr>
<td>Partition coefficient</td>
<td>The product contains mainly inorganic substances which are not biodegradable.</td>
</tr>
<tr>
<td>Auto-ignition temperature</td>
<td>Not relevant.</td>
</tr>
<tr>
<td>Decomposition Temperature</td>
<td>Not determined. During fire the product would decompose.</td>
</tr>
<tr>
<td>Viscosity</td>
<td>Not determined.</td>
</tr>
<tr>
<td>Explosive properties</td>
<td>Not explosive in its normal state. More sensitive to shock than m-dinitrobenzene: No. More sensitive to friction than m-dinitrobenzene: No. The mixture itself is not explosive but on reaction with metals can produce hydrogen gas which in sufficient concentration will explode.</td>
</tr>
</tbody>
</table>
HYDROCHLORIC ACID 0.1M

Explosive under the influence of a flame No

Oxidising properties Not relevant. The mixture is not an oxidising material.

Comments Not determined means the product was not tested for these properties. Not relevant means that these properties do not apply for this type of mixture due to its chemical properties.

9.2. Other information

Refractive index Not determined.
Particle size Not relevant.
Molecular weight Not relevant.
Volatile Not determined.
Saturation concentration Not determined.
Critical temperature Not relevant.
Volatil organic compound Not relevant.

SECTION 10: Stability and reactivity

10.1. Reactivity

Reactivity Can react with alkalis, oxidising agents and other acids. Reaction with sulphides can produce hydrogen sulphide gas. Reaction with cyanides can produce hydrogen cyanide gas. Reaction with metals can produce hydrogen gas which can form explosive atmospheres. Will corrode metals, some plastics and rubber. Ensure any packaging used to contain the mixture is compatible.

10.2. Chemical stability

Stability Stable when stored in sealed container at normal temperatures and in a suitable location.

10.3. Possibility of hazardous reactions

Possibility of hazardous reactions May react exothermically. May produce hydrogen cyanide or hydrogen sulphide. Reactions in a sealed container may result in pressure build up with possible rupture of the container. Will not polymerise.

10.4. Conditions to avoid

Conditions to avoid Avoid direct sunlight and moisture. Avoid heat and freezing conditions. Avoid storage with oxidising agents. Avoid storage with incompatible materials. It is advisable to store the product within some form of containment to prevent spillages reaching drainage systems. Avoid storage in an unstable manner or in a situation that would result in exposure to the product. Do not allow the storage container to be left exposed to the atmosphere.

10.5. Incompatible materials


10.6. Hazardous decomposition products

Hazardous decomposition products None anticipated at normal temperatures. See section 5 for thermal decomposition products.

SECTION 11: Toxicological information
HYDROCHLORIC ACID 0.1M

11.1. Information on toxicological effects

**Toxicological effects**
The mixture has not been tested for toxicological properties. Information regards hydrogen chloride or its mixtures. The mixture is not classified for toxic effects.

**Acute toxicity - oral**

*Notes (oral LD₅₀)*
Scientifically unjustified.

**Acute toxicity - dermal**

*Notes (dermal LD₅₀)*
Scientifically unjustified.

**Acute toxicity - inhalation**

**Acute toxicity inhalation (LC₅₀ vapours mg/l)**
8.3

*Notes (inhalation LC₅₀)*
30 minute exposure to HCl aerosol.

**Skin corrosion/irritation**

**Animal data**
Dose: 0.5ml of 170 g/l HCl in water, 4 hr, Rabbit OECD Guideline 404 Occlusive Corrosive 150 g/l not corrosive.

**Serious eye damage/irritation**

Tests on rabbits, OECD Guideline 405, Acute eye Irritation / Corrosion. HCl 5% w/w was found to be irritating to eyes (Category 1, irreversible effects on the eye).

**Respiratory sensitisation**

The mixture will irritate the respiratory system when vapours are inhaled, this may lead to sensitisation in certain individuals.

**Skin sensitisation**

Guinea pig maximization test (GPMT) - Guinea pig: OECD Guideline 406 (Skin sensitisation). Not sensitising.

**Germ cell mutagenicity**

Gene mutation:: Positive with metabolic activation. Mouse lymphoma cells.

**Genotoxicity - in vitro**

Scientifically unjustified. Industry - Dermal; Long term systemic effects 22 mg/kg/day:

**Carcinogenicity**

Not applicable. No carcinogenic effects. Not a carcinogen.

**Reproductive toxicity**

**Reproductive toxicity - fertility**
Scientifically unjustified. - Industry - Dermal; Long term systemic effects 22 mg/kg/day , Industry - Dermal; Long term systemic effects 22 mg/kg/day, Industry - Dermal; Long term systemic effects 22 mg/kg/day

**Reproductive toxicity - development**
No information available. Industry - Dermal; Long term systemic effects 22 mg/kg/day - Industry - Dermal; Long term systemic effects 22 mg/kg/day: Industry - Dermal; Long term systemic effects 22 mg/kg/day, Industry - Dermal; Long term systemic effects 22 mg/kg/day, Industry - Dermal; Long term systemic effects 22 mg/kg/day, Industry - Dermal; Long term systemic effects 22 mg/kg/day No reliable information.

**Specific target organ toxicity - single exposure**

**STOT - single exposure**
These effects will be noticed when vapours are produced and inhaled by exposed operators. Respiratory irritant effects that impair function with symptoms such as cough, pain, choking, and breathing difficulties.

**Target organs**
Respiratory system, lungs

**Specific target organ toxicity - repeated exposure**

**STOT - repeated exposure**
LOAEL 50 ppmV/6hr/day, Inhalation, Mouse 90 day exposure.
HYDROCHLORIC ACID 0.1M

Target organs
Respiratory system, lungs

General information
Effects will be dependent upon the concentration and length of time of exposure. Higher concentrations will produce more pronounced effects.

Inhalation
Gas or vapour in high concentrations may irritate the respiratory system. Symptoms following overexposure may include the following: Coughing.

Ingestion
May irritate mouth, throat and gastrointestinal tract. Ingestion of large amounts may cause nausea and vomiting.

Skin contact
May irritate the skin.

Eye contact
May irritate the eyes.

Medical symptoms
Irritation of the eyes, respiratory system and skin. Coughing and difficulties with breathing. Irritation of mouth, throat and oesophagus. Nausea, vomiting.

SECTION 12: Ecological Information

Ecotoxicity
The product may affect the acidity (pH) of water which may have hazardous effects on aquatic organisms. The mixture has not been tested for ecotoxicological properties, the information refers to hydrogen chloride or its mixtures.

12.1. Toxicity

Acute toxicity - fish
LC50, 96 hours: pH 3.25 - 3.5 Industry - Dermal; Long term systemic effects 22 mg/kg/day, Lepomis macrochirus (Bluegill)
Freshwater, semi-static.

Acute toxicity - aquatic invertebrates
EC50, 48 hours: pH 4.92 , Daphnia magna
Static, freshwater.

Acute toxicity - aquatic plants
EC50, 72 hours: pH 4.7 , OECD Guideline 201 (Algae Growth Inhibition Test) peformed on Chlorella Vulgaris.
Static, freshwater.

Acute toxicity - microorganisms
EC50, 3 hours: pH 5.0 - 5.5 , Activated sludge
OECD Guideline 209: Activated Sludge, Respiration Inhibition Test.
Static, freshwater.

Acute toxicity - terrestrial
Not available.
No supplied or registered information

Chronic toxicity - fish early life stage
Not available.
No supplied or registered information

Short term toxicity - embryo and sac fry stages
Not available.
No supplied or registered information

Chronic toxicity - aquatic invertebrates
Scientifically unjustified.

Toxicity to soil
No registered or supplied information.

Toxicity to terrestrial plants
No registered or supplied information.

12.2. Persistence and degradability

Persistence and degradability
The product is not biodegradable. Hydrochloric acid dissociates completely in water and soil to form chloride ions and hydroxonium ions. Minerals in the soil will help to neutralise the acid.
HYDROCHLORIC ACID 0.1M

Phototransformation
Not relevant.
Hydrochloric acid is an inorganic compound with no nitrogen groups, hydroxide groups, double bonds, triple bonds or aromatic rings. It dissociates readily when dissolved in atmospheric moisture therefore an estimation of phototransformation is not practical.

Stability (hydrolysis)
Not relevant.
Hydrochloric acid is not hydrolysed.

Biodegradation
Scientifically unjustified.
Hydrochloric acid is not biodegradable as it dissociates in contact with water and soil water.

Biological oxygen demand
Not relevant.

Chemical oxygen demand
Not relevant.

12.3. Bioaccumulative potential
Bioaccumulative potential
The product is not bioaccumulating. Study scientifically unjustifiable.

Partition coefficient
The product contains mainly inorganic substances which are not biodegradable.

12.4. Mobility in soil
Mobility
Minerals in the soil tend to neutralise acid contamination however larger or continuous emissions may lead to the product travelling into groundwater. As the product travels further into the soil the increased contact raises the pH to make it less acidic.

Adsorption/desorption coefficient
Scientifically unjustified. Test methods are not applicable for molecules that dissociate. Following dissociation ions are expected to undergo ion exchange with the soil.

Henry's law constant
Not determined.

Surface tension
Not available. No supplied or registered information

12.5. Results of PBT and vPvB assessment
Results of PBT and vPvB assessment
This product does not contain any substances classified as PBT or vPvB.

12.6. Other adverse effects
Other adverse effects
Reaction with hypochlorites can produce chlorine gas. Will affect drinking water supplies. May cause a local pH change in water systems which can affect aquatic organisms. May effect germination and growth rates of plants if soil contamination occurs.

SECTION 13: Disposal considerations

13.1. Waste treatment methods
General information
The mixture is acidic and classed as corrosive to metals, therefore it should be treated as hazardous waste. Do not dispose to drain without prior consultation with local authorities. If operators are exposed to vapours during the disposal process then suitable respiratory protection should be worn. All other personal protective equipment as described in section 8 should be worn.
HYDROCHLORIC ACID 0.1M

Disposal methods
Dispose of waste to licensed waste disposal site in accordance with the requirements of the local Waste Disposal Authority. Uncleaned empty containers should be treated as hazardous waste. Neutralisation is recommended before disposal, this should be carried out by an approved waste treatment company. When the material is to be treated on site, this should be done with an aqueous slurry of sodium carbonate. BE AWARE THAT HEAT WILL BE GENERATED AND A VIOLENT REACTION MAY OCCUR. ALWAYS WEAR PROTECTIVE CLOTHING AND EQUIPMENT AS SPECIFIED IN SECTION 8. Waste material should not be disposed of directly to drain. Avoid unauthorised disposal. Do not dump illegally onto land or into water. When dealing with waste always consider the waste management hierarchy of Prevention, Preparation for re-use, Recycling, Recovery and Disposal. It is advisable to minimise waste at source if possible, then re-use, recover or recycle wherever possible before considering waste disposal options.

SECTION 14: Transport information

14.1. UN number
UN No. (ADR/RID) 1789
UN No. (IMDG) 1789
UN No. (ICAO) 1789

14.2. UN proper shipping name
Proper shipping name (ADR/RID) HYDROCHLORIC ACID
Proper shipping name (IMDG) HYDROCHLORIC ACID
Proper shipping name (ICAO) HYDROCHLORIC ACID
Proper shipping name (ADN) HYDROCHLORIC ACID

14.3. Transport hazard class(es)
ADR/RID class 8
ADR/RID label 8
IMDG class 8
ICAO class/division 8

Transport labels

14.4. Packing group
ADR/RID packing group III
IMDG packing group III
ICAO packing group III

14.5. Environmental hazards
Environmentally hazardous substance/marine pollutant No.

14.6. Special precautions for user
EmS F-A, S-B
HYDROCHLORIC ACID 0.1M

Emergency Action Code 2R
Hazard Identification Number (ADR/RID) 80
Tunnel restriction code (E)

14.7. Transport in bulk according to Annex II of MARPOL and the IBC Code
Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code Not applicable.

SECTION 15: Regulatory information

15.1. Safety, health and environmental regulations/legislation specific for the substance or mixture
Control of Substances Hazardous to Health Regulations 2002 (as amended).
Guidance Workplace Exposure Limits EH40.
Industry - Dermal; Long term systemic effects 22 mg/kg/day
ECHA Guidance on the compilation of safety data sheets 2014.

15.2. Chemical safety assessment
No chemical safety assessment has been carried out. Information from the manufacturer of the raw material has not been received regarding Chemical Safety Assessments, Exposure Scenarios or a Chemical Safety Report.

SECTION 16: Other information

General information This datasheet is not intended to be a replacement for a full risk assessment, these should always be carried out by competent persons. Toxicological and ecotoxicological information has been taken from the ECHA website of registered substances.

Key literature references and sources for data ECHA website. Health Protection Agency Information. Raw material safety data sheets.

Revision comments Change to section 15
Revision date 28/04/2016
Revision 3
Supersedes date 30/07/2015
SDS number 11781
Risk phrases in full Not classified.
R34 Causes burns.
R37 Irritating to respiratory system.
HYDROCHLORIC ACID 0.1M

Hazard statements in full

H290 May be corrosive to metals.
H314 Causes severe skin burns and eye damage.
H318 Causes serious eye damage.
H335 May cause respiratory irritation.